National 5 Assignment
Latent heat of vaporisation Guide Sheet: A

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**To determine the specific latent heat of vaporisation of water.**

*This experiment is only suitable if a graph can be plotted.*

**Apparatus**

Scales, electric jug kettle with automatic cut-off overridden, stopclock

**Instructions:**



* Set up the apparatus as shown.
* Note the power rating of the kettle.
* Switch the kettle on and allow the water to boil; at this point note the balance reading.
* Start the stopclock as the water begins to boil.
* Take readings of the water boiled against the time from the start of boiling
* Stop the clock after 0.2 kg of water has boiled off.
* Calculate the energy required to evaporate each quantity of water.
* Use your data to determine the specific latent heat of vaporisation of water.
* The energy required to evaporate 1 kg of boiling water is the specific latent heat of vaporisation of water.

**Risk Assessment**

* Safety googles must be worn.
* Electrical wiring must be checked.
* Do not move the kettle once set up.
* Have access to the plug so that the electrical supply can be switched off
* ***Check with SSERC that this experiment is suitable. If mains is being used as the heater it might not be appropriate.***
* The kettle can be replaced with a heater plugged into a joulemeter and the energy recorded every few minutes and the mass lost recorded.